



SAFETY LEAD MATERIAL SELECTION

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30 June 1976

INTRODUCTION

The Energy Doubler superconducting magnets in their present design require safety leads for prompt extraction of the stored energy in the event of a quench. The design criteria for these leads differ from the one for the power leads since the safety lead will carry current only during infrequent and short periods.

In this memo, the problem is analyzed, a key parameter involved (the integral of the specific heat over the resistivity from room temperature to a still safe temperature T_{\max}) is calculated for several materials using solid state theory and Debye Temperature values from the literature. This parameter is useful when comparing the capacity of metals for handling current surges. The computer programs developed to do these calculations are included as an appendix.

ANALYSIS

A conservative design criterion is that the energy deposited in any section of the lead be insufficient to raise its temperature above the safe value T_{\max} under the worst external cooling conditions; namely, no external cooling.

Let A , ρ , and c be the cross section area, resistivity and specific heat of an element of length dx of the lead. The electrical resistance of this element will be $R = \rho(T) \frac{dx}{A}$ and its

heat capacity $C = \mu A c(T) dx$ where μ is the density of the material. When a time dependent current $I = I(t)$ goes through the lead, the energy deposited in the lead per unit time by Joule heating raises the temperature, T , of the lead element according to the energy balance expression:

$$A \mu c(T) \frac{dT}{dt} dx = \left(\frac{1}{A}\right) I^2(t) \rho(T) dx. \quad (1)$$

According to the above mentioned criterion we are neglecting heat transfer into or away from this element of lead. The usual short duration, Δt , of $I(t) \neq 0$ validates this approximation which in any case represents the worse case from a safety point of view.

Further simplification of the problem is made by assuming a uniform cross section lead, which is the preferred type from the standpoint of fabrication.

Expression (1) in an integral form reads:

$$A^2 \mu \int_{T_{\text{initial}}}^{T_{\text{max}}} \frac{c(T)}{\rho(T)} dT = \int_0^{\Delta t} I^2(t) dt. \quad (2)$$

The right hand side could be called a "quench load", and is expressed in $A^2 \cdot \text{sec}$ or Joules/Ohms. In fuse literature it is known by the misleading name of fuseing current. A maximum $I(t)$ expected in the quench of an Energy Doubler dipole is graphically shown in Fig. 1. Its corresponding quench load is:

$$F = \int_0^{\Delta t} I^2(t) dt = 6.66 \times 10^6 \text{ Joule}/\Omega.$$

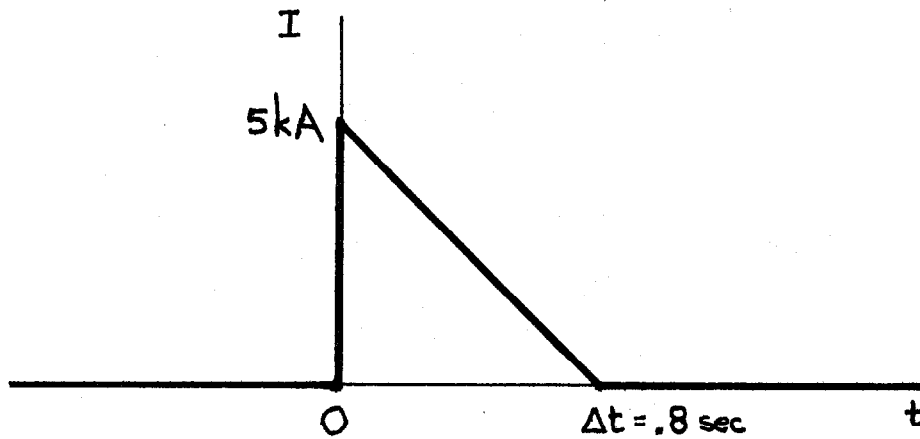


Figure 1

Maximum Current Pulse Expected From a Quench
of an Energy Doubler Dipole

QUENCH CAPABILITY

The left hand side of expression (2) depends only on the lead material, its cross section and the temperature limits. By specifying the initial temperature as room temperature (the highest temperature before the current surge) and T_{\max} as the highest safe temperature, we define the "quench capability", f , of the lead material as:

$$f = \mu \int_{300K}^{T_{\max}} \frac{c(T)}{\rho(T)} dT.$$

According to solid state theory both the specific heat and the resistivity as a function of temperature can be expressed in terms of the Debye Temperature, θ , of the material and its resistivity at this temperature, $\rho(\theta)$.

The specific heat of a solid near and above room temperature is mainly due to lattice vibrations and according to the Debye theory¹ is given by

$$c(T) = \frac{9k_B N}{M} \left(\frac{T}{\Theta} \right)^3 \int_0^{\Theta/T} \frac{e^Y y^4 dy}{(e^Y - 1)^2}$$

where k_B = Boltzmann constant,

N = Avogadro's number,

and M = Molecular weight of material.

Also the resistivity can be expressed in terms of a dimensionless integral² when the electrons are scattered mainly by phonons:

$$\rho(T) = \rho(\Theta) \left(\frac{T}{\Theta} \right)^5 \int_0^{\Theta/T} \frac{y dy}{(e^Y - 1)(1 - e^{-Y})}$$

One of the computer programs in the appendix, SL3.F4, calculates the above integrals and their ratio to within .2%, generating a table SL4.DAT which is used in the second program SL4.F4. This second program requests data on the material (μ , M , Θ , $\rho(\Theta)$) and the thermal conductivity integral $\int_{4.2K}^{300K} \lambda dT$ and calculates the quench capability, f , as well as the merit figure z explained below.

When a safety lead is not cooled by exchanging heat with a gas flow, the conduction heat load, Q , should be minimized.

$$\text{From } Q = \frac{A}{\ell} \int_{4.2K}^{300K} \lambda dT \text{ where } \lambda \text{ is the thermal conductivity and}$$

l is the length of the lead, and $F = A^2 \cdot f$ we get

$$Q = \sqrt{\frac{F}{f}} \frac{1}{l} \int_{4.2K}^{300K} \lambda dT .$$

For a fixed design, the material with largest merit figure:

$$z = \sqrt{f} / \int_{4.2K}^{300K} \lambda dT$$

should be preferred if all other mechanical and economic constraints are satisfied.

However, vapor cooled leads are very efficient in reducing the conduction heat load and a material with largest f compatible with other fabrication constraints should be selected.

On Table I, the parameters f and z are presented for a list of single element materials in the case of a heating excursion from room temperature to 180°C (soft solder lowest melting point is 185°C). The material properties used as input were obtained from the literature³, which in many cases does not have unique values for θ or the thermal conductivity integral. Average values were then used and the percental error indicated in f covers the spread found in the literature.

On Fig. 2, the quench capability of copper and nickel are shown as a function of the upper limit of the heating excursion. The third program of the Appendix, SL5.F4, calculates this result.

CONCLUSION

The materials examined classify themselves with respect to quench capability inversely to their resistivity at the Debye Temperature, and copper seems an easy choice when the conduction heating is compensated by vapor cooling. Lead is an exception.

For leads with uniform cross section and without vapor cooling niobium followed by nickel are better choices since lead might be too soft.

Copper plated stainless steel, might be the best engineering compromise for simple vapor cooled leads.

ACKNOWLEDGEMENT

The interest and cooperation of R. H. Flora in this work was sincerely appreciated.

REFERENCES

¹ See for instance "Physics of Electronic Conduction in Solids" by Frank J. Blatt, p. 46, McGraw Hill Book Company (1968).

² Ibid, p. 189.

³ Ibid, p. 48 and 192 for θ ; Handbook of Chemistry and Physics for μ and M ; L. A. Hall, NBS Technical Note 365 for $\rho(\theta)$; R. B. Stewart and V. J. Johnson, WADD Technical Report 60-56 for

$$\int_{4.2K}^{300K} \lambda dT$$

TABLE I

Quench Capability and Merit Figure of Several Elements
For A Heating Excursion From 300 To 453K

<u>Material</u>	<u>f</u> <u>MJ/Ωcm^4</u>	<u>z</u> <u>sec $^{1/2}$ cm$^{-1}$ V$^{-1}$</u>
Au	630. \pm 5%	.022
Ag	557. \pm 2%	.010
Cu	319. \pm 5%	.012
Pb	143. \pm 14%	.075
Al	96.2 \pm 1%	.0063
W	89.5 \pm 7%	.004
Mo	48.6 \pm 9%	.014
Ta	46.6 \pm 8%	.038
Ni	44.6	.031
Nb	41.5 \pm 5%	.041
Fe	31.7 \pm 5%	.014

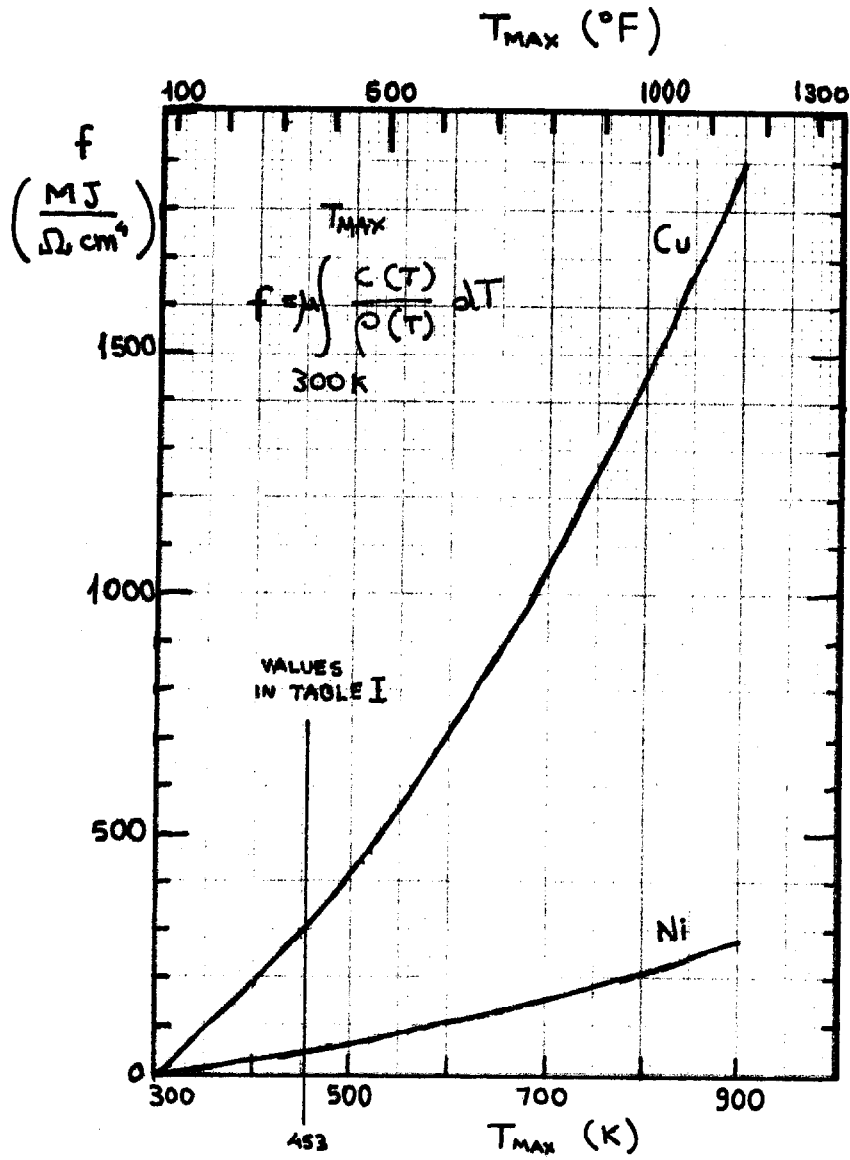


Figure 2

Quench Capability of Copper and Nickel

APPENDIX

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00100 C      SAFETY LEAD MATERIAL SELECTION SCHEME
00200 C
00300 C SL3:  BASED IN THE DEBYE THEORY THIS PROGRAM GENERATES TABLES OF
00400 C          SPECIFIC HEAT, RESISTIVITY AND THEIR RATIO IN A DIMENSIONLESS
00500 C          FORM. ITS OUTPUT, SL4.DAT, IS TO BE USED BY SL4.F4.
00600 C          EXECUTION REQUEST:  EXEC SL3,DSKB:QATR[10,6]
00700 C          PRECISION OF RATIO: .2%
00800 C
00900      DIMENSION C(2500),R(2500),CR(2500),AUX(100)
01000      EXTERNAL SH,RHO
01100      XL=0.
01200      EPS=5.E-6
01300      EPSR=1.E-6
01400      NDIC=5
01500      NDIR=5
01600      IER=3
01700      Y=0.
01800      IA=0
01900      CALL OFILE(20,'SL4.DAT')
02000      WRITE(20,6)
02100 C GENERATE TABLES OF INTERNAL INTEGRALS AND RATIO
02200      DO 1 I=150,2200,1
02300 C      REPORT CALCULATION PROGRESS TO TERMINAL
02400      AI=I
02500      II=AI/10.
02600      IF (II.NE.IA) WRITE(5,7) I
02700      IA=II
02800 7      FORMAT(' ',I4)
02900      XI=I
03000      XU=XI/1000.
03100      J=I-149
03200 31      CALL QATR(XL,XU,EPSC,NDIC,SH,Y,IER,AUX)
03300      EPSC=1.E-3*Y
03400      C(J)=Y
03500      IF (IER.EQ.2) GO TO 30
03600      IF (IER.NE.0) WRITE(5,2) IER,XU,Y,NDIC
03700 2      FORMAT(' QATR ERROR',I2,' SH(',F6.3,')=',F,I)
03800 33      CALL QATR(XL,XU,EPSR,NDIR,RHO,Y,IER,AUX)
03900      EPSR=1.E-3*Y
04000      R(J)=Y
04100      IF (IER.EQ.2) GO TO 32
04200      IF (IER.NE.0) WRITE(5,3) IER,XU,Y,NDIR
04300 3      FORMAT(' QATR ERROR',I2,' RHO(',F6.3,')=',F,I)
04400      CR(J)=C(J)/R(J)
04500      WRITE(20,5) XU,C(J),R(J),CR(J)
04600 1      CONTINUE
04700 C PRINT SUMMARY TABLE
04800      WRITE(5,6)
04900 6      FORMAT('          TD/T          C/ATOM RESISTIVITY          C/R')
05000      DO 4 I= 200,1500,100

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05100      XI=I
05200      X=XI/1000.
05300      J=I-149
05400      WRITE(5,5) X,C(J),R(J),CR(J)
05500      5  FORMAT(' ',F11.4,2E11.3,F11.4)
05600      4  CONTINUE
05700      STOP
05800      C INCREASE QATR PRECISION
05900      30  IF(NDIC.GE.100) GOTO 31
06000      NDIC=NDIC+1
06100      WRITE(5,34)NDIC,XU
06200      34  FORMAT(' NDIC INCREASED TO',I3,' AT X=',F)
06300      GO TO 31
06400      32  IF(NDIR.GE.100) GOTO 33
06500      NDIR=NDIR+1
06600      WRITE(5,35)NDIR,XU
06700      35  FORMAT(' NDIR INCREASED TO',I3,' AT X=',F)
06800      GOTO 33
06900      END
07000      C DEBYE THEORY FUNCTIONS
07100      FUNCTION SH(X)
07200      SH=0.
07300      IF(X.EQ.0.)RETURN
07400      Y=EXP(X)
07500      SH=Y♦X♦♦4/((Y-1.)♦♦2)
07600      RETURN
07700      END
07800      FUNCTION RHO(X)
07900      SH=0.
08000      IF(X.EQ.0.)RETURN
08100      RHO=X♦♦5/((EXP(X)-1.)♦(1.-EXP(-X)))
08200      RETURN
08300      END

```

PRINTOUT OF SUMMARY TABLE

2190
2200

TD/T	C/ATOM	RESISTIVITY	C/R
0.2000	0.266E-02	0.399E-03	6.6642
0.3000	0.896E-02	0.202E-02	4.4440
0.4000	0.212E-01	0.635E-02	3.3343
0.5000	0.412E-01	0.154E-01	2.6688
0.6000	0.707E-01	0.318E-01	2.2254
0.7000	0.112E+00	0.585E-01	1.9088
0.8000	0.165E+00	0.989E-01	1.6716
0.9000	0.233E+00	0.157E+00	1.4873
1.0000	0.317E+00	0.237E+00	1.3400
1.1000	0.418E+00	0.343E+00	1.2196
1.2000	0.537E+00	0.479E+00	1.1194
1.3000	0.674E+00	0.651E+00	1.0347
1.4000	0.831E+00	0.864E+00	0.9622
1.5000	0.101E+01	0.112E+01	0.8995

(Table SL4.DAT
available on
request)

END OF EXECUTION
CPU TIME: 56:22.95 ELAPSED TIME: 2:31:46.83
EXIT

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00100 C SAFETY LEAD MATERIAL SELECTION SCHEME
00200 C
00300 C SL4: THIS PROGRAM USES THE OUTPUT TABLE SL4.DAT
00400 C GENERATED BY SL3.F4, REQUESTS MATERIAL DATA AND TEMPERATURE
00500 C LIMITS AND CALCULATES THE MATERIAL DEPENDENT
00600 C INTEGRAL F AND THE MERIT INDEX Z
00700 C
00800 DIMENSION C(2500),R(2500),CR(2500)
00900 CALL IFILE(1,'SL4.DAT')
01000 READ(1,6)
01100 6 FORMAT('
01200 DO 1 I=150,2200,1
01300 J=I-149
01400 READ(1,5)XU,C(J),R(J),CR(J)
01500 5 FORMAT(' ',F11.4,2E11.3,F11.4)
01600 1 CONTINUE
01700 9 WRITE(5,7)
01800 7 FORMAT(' ENTER TI AND TM (INITIAL AND MAXIMUM TEMP (K))')
01900 READ(5,8)TI, TM
02000 8 FORMAT(26)
02100 FLAG=0.
02200 IF (TI.EQ.0) STOP
02300 15 IF (FLAG.EQ.1.) GOTO 3
02400 WRITE(5,11)
02500 11 FORMAT(' ENTER: ATOMIC WEIGHT(G), DENSITY(G/CM3), TDEBYE(K),'''
02600 1 RESISTIVITY(E-6.OHM.CM) AT TDEBYE,''' THERMAL
02700 2CONDUCTIVITY INTEGRAL (WATT/CM) FROM 4 TO 300K. '''
02800 3 (FOLLOWING PROMPTERS WILL JUST ASK FOR: A,D,TD,RTD,TCI)')
02900 FLAG=1.
03000 16 READ(5,12) A,D,TD,RTD,TCI
03100 12 FORMAT(5G)
03200 13 IF (A.EQ.0) GOTO 9
03300 13 GG=17.706*D*TD/(A*RTD)
03400 XL=TD/TM
03500 XU=TD/TI
03600 WRITE(5,2)XL,XU
03700 2 FORMAT(' XL=',F11.4,' XU=',F11.
03800 IF (XL.LT..15.OR.XU.GT.2.2) GOTO 9
03900 LL=XL*1000.-149
04000 LU=XU*1000.-148
04100 H=.001*.5
04200 K=LL
04300 C INTEGRATION LOOP
04400 SUM=0.
04500 10 K=K+1
04600 X22=K+149
04700 X22=(1000./X22)**2
04800 X21=K+148
04900 X21=(1000./X21)**2
05000 SUM1=SUM
05100 SUM=SUM+H*(CR(K)*X22+CR(K-1)*X21)
05200 IF (K.LT.LU) GOTO 10
05300 F=GG*SUM
05400 Z=SQRT(F)/TCI
05500 WRITE(5,14) F,Z
05600 14 FORMAT(' QUENCH CAPABILITY=',F6.2,' JOULES/(OHM*CM4) MERIT
05700 1 VALUE=',F6.4)
05800 GOTO 15
05900 3 WRITE(5,4)
06000 4 FORMAT(' ENTER: A,D,TD,RTD,TCI')
06100 GO TO 16
06200 FMT
06300

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00100  C      SAFETY LEAD MATERIAL SELECTION SCHEME
00200  C
00300  C  SL5:  THIS PROGRAM USES THE OUTPUT TABLE SL4.DAT GENERATED BY SL3.F4
00400  C      FOR TI=300.K AND GENERATES F(TM).
00500  C      MATERIAL PARAMETERS ARE REQUESTED LIKE IN SL4 FAST PROMPTER.
00600  C      ZERO ENTRY TERMINATES RUN.
00700  C
00800      DIMENSION C(2500),R(2500),CR(2500)
00900      CALL IFILE(1,'SL4.DAT')
01000      READ(1,6)
01100  6      FORMAT('
01200      DO 1 I=150,2200,1
01300      J=I-149
01400      READ(1,5)XU,C(J),R(J),CR(J)
01500  5      FORMAT(' ',F11.4,2E11.3,F11.4)
01600  1      CONTINUE
01700      A=63.5
01800      D=8.96
01900      TD=343.
02000      RTD=1.85
02100      TCI=1500.
02200  3      WRITE(5,4)
02300  4      FORMAT(' ENTER: A, D, TD, RTD, TCI')
02400      READ(5,12)A,D,TD,RTD,TCI
02500  12      FORMAT(5G)
02600      IF (A.EQ.0.) STOP
02700      TI=300.
02800      WRITE(5,7)
02900  7      FORMAT(' TI=300.K// TM (K) F (MJ/DHM.CM4) Z')
03000      DO 30 KK=310,900,10
03100      TM=KK
03200      GG=17.706*D*TD/(A*RTD)
03300      XL=TD/TM
03400      XU=TD/TI
03500      IF(XL.LT..15.OR.XU.GT.2.2) GOTO 9
03600      LL=XL*1000.-149
03700      LU=XU*1000.-148
03800      H=.001*.5
03900      K=LL
04000  C  INTEGRATION LOOP
04100      SUM=0.
04200  10      K=K+1
04300      X22=K+149
04400      X22=(1000./X22)**2
04500      X21=K+148
04600      X21=(1000./X21)**2
04700      SUM1=SUM
04800      SUM=SUM+H*(CR(K)*X22+CR(K-1)*X21)
04900      IF(K.LT.LU) GOTO 10
05000      F=GG*SUM
05100      Z=SQRT(F)/TCI
05200      WRITE(5,14) TM,F,Z
05300  14      FORMAT(' ',F5.0,F12.2,F13.4)
05400  30      CONTINUE
05500      GOTO 3
05600  9      WRITE(5,2)XL,XU
05700  2      FORMAT(' XL=',F11.4,' XU=',F11.4)
05800      STOP
05900      END

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